Synthesis, Characterization and Biological Activityof some novelphenyl-2H, 1'H-[3, 4'] bipyrazolyl derivatives.

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Abstract

An important class of heterocyclic molecule called bi-pyrazolyl and its derivatives has both chemical and biological applications. In this study, substituted acetophenones and substituted hydrazine are used to create bi-pyrazolyl derivatives, which are then synthesised and characterised. Substituted acetophenones, substituted hydrazine, and DMF were used in the first phase of the synthesis of 1-substituted-3-aryl-1H-pyrazole-4-carbaldehydes derivatives in phosphorus trichloride (POCl₃) using a scientific microwave oven 20%, 140 watts. In the second phase, various phenyl-2H, 1'H-[3, 4] bipyrazolyl derivatives with good yield were produced by reacting substituted hydrazine, acetone, and 10% NaOH with 1-substituted 3-aryl-1H-pyrazole-4-carbaldehydes derivatives in the scientific microwave oven (20%, 140 watts). FT-IR, H1NMR, and 13CNMR were used to characterise each of the produced compounds. By using PASS, all of the produced compounds were examined for biological activity.

Keywords:-DMF, substituted hydrazine, phosphorus trichloride, QSAR study, microwave irradiation.

I. Introduction

The first bipyrazolyl series was created in 1893¹. Since then, numerous articles about bipyrazolyl derivatives have been published in the literature ². These substances represent a very exceptional class of pharmacologically relevant heterocycles. They may have antitumor³, anti-inflammatory⁴, antimicrobial, cytotoxic^{5, 6}, antiallergic⁷, cardiovascular⁸, and diuretic⁹ properties, for instance. As insecticides¹⁰, herbicides¹¹, fungicides^{12–14}, in the photography and paint industries¹⁵, and in the manufacture of heat-resistant polymers, bipyrazoles were also discovered to be useful. Additionally, bipyrazole derivatives have been employed as free radical scavengers¹⁶ and to prevent or treat a number of conditions brought on by active oxygen.

Bipyrazolyl derivatives are often created using cyclization procedures¹⁷. However, 1, 3dipolar cycloaddition¹⁸, which uses diarylnitrilimines or diazocompounds as reagents, is one of the most flexible methods for the synthesis of five-membered heterocycles. These methods have also been utilised to produce bipyrazoles. Since 1978, it has been known that hydrazones can form azomethine imine intermediates that can go through a 1,3-dipolar cycloaddition by inducing a thermal 1,2-hydrogen shift. A protonated substrate has been shown to intramolecularly aid the hydrazone-azomethine imine isomerization under benign conditions¹⁹ if a carbonyl group is present in the proper location.However, in the majority of instances, high-activated dipolarophiles must be added to thermally generated azomethineimines under reflux in high-boiling solvents with extended reaction durations²⁰. The reaction's synthetic usefulness is decreased by these requirements.

Considering the significance of bipyrazolyl derivatives, we aimed to create novel substances like as 2-(2, 4-Dinitro-phenyl)-5-methyl-3'-phenyl-2H,1'H-[3,4']bipyrazolyl[5-a], 3'-(4-Methoxy-phenyl)-5-methyl-2-5-Methyl-2,3'-diphenyl-2H,1'H-[3,4']bipyrazolyl[5-b], phenyl-2H,1'H-[3,4']bipyrazolyl [5-c], 2-(2,4-Dinitro-phenyl)-3'-(4-methoxy-phenyl)-5methyl-2H,1'H-[3,4']bipyrazolyl 3'-(4-Methoxy-phenyl)-5-methyl-2,1'-diphenyl-[5-d], 2H,1'H-[3,4']bipyrazolyl [5-e], 2-(2,4-Dinitro-phenyl)-3'-(4-methoxy-phenyl)-5-methyl-1'phenyl-2H,1'H-[3,4']bipyrazolyl 5-Methyl-3'-(4-nitro-phenyl)-2-phenyl-2H,1'H-[5-f], [3,4']bipyrazolyl 2-(2,4-Dinitro-phenyl)-5-methyl-3'-(4-nitro-phenyl)-2H,1'H-[5-g], [3,4']bipyrazolyl[5-h], 5-Methyl-3'-(4-nitro-phenyl)-2,1'-diphenyl-2H,1'H-[3,4']bipyrazolyl 2-(2,4-Dinitro-phenyl)-5-methyl-3'-(4-nitro-phenyl)-1'-phenyl-2H,1'H-[5-i] and [3,4']bipyrazolyl [5-j].

PASS

A user-selected (set of) compound's 565 likely biological functions are predicted by the software programme PASS (s). 5-hydroxytryptamine antagonists, neuromuscular blockers, antibiotics, antidepressants, antiviral medicines, contraceptives, tumour necrosis factor antagonists, and many other substances are included in these actions. PASS predictions have led to the discovery of novel pharmaceutical substances with impacts on anxiety, inflammation, hypertension, cancer, and other conditions. Chemical libraries of millions of chemicals are covered by PASS.

According to various functions, as illustrated in equation, the biological activities of chemical compounds are related to their physical characteristics (1).

Biological activity = f (physicochemical properties).....(1)

As a result, "the biological activity spectrum" is described as a quality that is "intrinsic" to a material and solely dependent on its physicochemical properties. Based on SAR analysis of the training set, which contains hundreds of chemicals with various biological functions, PASS is able to predict this spectrum. The qualitative descriptions of biological activities are included in PASS.

Importance of PASS

1. Since it takes time and money to experimentally determine a drug's biological function, using PASS is frequently important.

2. PASS is a useful tool for finding compounds with the desired qualities and no undesired side effects.

3. It is used to select the compounds that are the most likely from the pool of samples that are available for a particular screening.

4. To identify further relevant screens for a specific component.

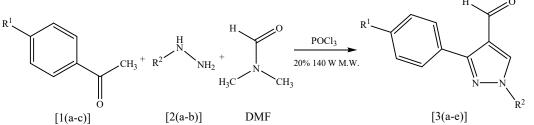
The current study uses PASS as a technique to create the medicine with the highest potential activity because of the significance of PASS.

II. Experimental

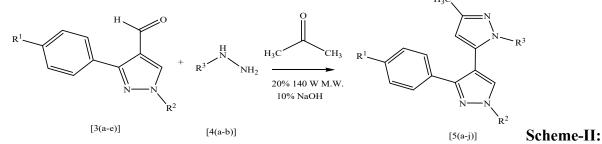
The melting points were discovered in uncorrected open capillary tubes. The 1H NMR spectra were taken in CDCl₃ with TMS as an internal standard on a Bruker spectrometer at 400 MHz, and the IR spectra were captured in KBr pellets on a Nicolet 400D spectrometer. Selected samples collected with the LC-MSD-Trap-SL 01046 and LC-MS TLC was used to test the compounds' purity on silica-G plates. Using pass software, anti-microbial activities are checked.

Synthesisof1-substituted-3-aryl-1H-pyrazole-4-carbaldehydes(3a-e): In 100 mL round bottom flask 0.02 mol substituted acetophenone(1a-c), 0.02 mol substituted hydrazine (2a-b) and 10 mL DMF were mixed in ethanol (30 mL). POCl₃ solution (1 mL) was added and microwaves were used to irradiate the reaction mixture for 4-5 minutes at 20% microwave power (140 W).TLC was used to confirm the reaction's completion (Ethyl acetate: Hexane 1:9). The reaction mixture was put into crushed ice after being allowed to cool at room temperature. To obtain pure products, the solid was filtered, washed with a little amount of ethanol, then refined by recrystallization from ethanol (3a-e)(Scheme-I).

Synthesisof phenyl-2H, 1'H-[3, 4'] bipyrazolyl (5a-j): A 100 mL round bottom flask 0.001 mol3-Phenyl-1H-pyrazole-4-carbaldehyde **(3a-e)**, 0.001 mol substituted hydrazine **(4a-b)**, 10 mL acetone were mixed in ethanol (30 mL) and 10% NaOH solution were added. Microwaves were used to irradiate the reaction mixture for 5-7 minutes at 20% microwave power (140 W). TLC was used to confirm the reaction's completion (Ethyl acetate: Hexane 1:9). The reaction mixture was put into crushed ice after being allowed to cool at room temperature. To obtain pure products, the solid was filtered, washed with a little amount of ethanol, then refined by recrystallization from ethanol **(5a-j)(Scheme-II)**. Table 1 summarises the specifics of the microwave-induced bipyrazolyl synthesis.



Scheme-I: Synthesisof1-substituted-3-aryl-1H-pyrazole-4- carbaldehydes



Synthesisofphenyl-2H, 1'H-[3, 4'] bipyrazolyl

2-(2, 4-Dinitro-phenyl)-5-methyl-3'-phenyl-2H, 1'H-[3, 4']bipyrazolyl (5a): Yield: 83%; Time (min.): 5; m.p.214⁰C;m.f. C₁₉H₁₄N₆O₄;IR (KBr, ν_{max}, cm⁻¹): 1124.14 (C-N), 1241.86 (N-N), 1492.90 (-NO₂), 1583.56 (C=N), 3095.75 (Ar-CH), 3309.85 (NH);¹HNMR (δ ppm): 2.33 (s, 3H, CH₃), 6.45 (s, 1H, =CH), 7.42-7.44 (d, 1H,=CH), 7.61-7.64 (m, 5H, C₆H₅), 7.81-7.84 (d, 1H, =CH), 8.44-8.47 (d, 1H, =CH), 8.75 (s,1H, =CH), 9.52 (s, 1H, NH); ¹³C NMR(δ ppm): 13.50(-CH₃), 108.03(=CH), 115.16(=C<),121.58(=CH), 126.58(2×=CH), 127.58(=CH), 128.20(=CH),129.00(2×=CH), 130.50(=CH), 136.34(=C<), 137.17(=C<), 138.60(=CH), 142.34(=C<),144.30(=C<), 145.20 (=C<), 146.58(=C<), 150.00(=C<); MS:*m/z* (M⁺1 390).

5-Methyl-2, 3'-diphenyl-2H, 1'H-[3, 4'] bipyrazolyl (5b): Yield: 76%; Time (min.): 5; m.p. 209^{0} C; m.f. $C_{19}H_{16}N_{4}$; IR (KBr, v_{max} , cm⁻¹):1075.24(C-N), 1217.24(N-N), 1504.20(C=C), 1600.35(C=N), 3021.24(Ar-CH), 3300.56(NH) ; ¹HNMR (δ ppm): 2.33(s, 3H, CH₃), 6.45(s, 1H, =CH), 7.37-7.40(m, 5H, C₆H₅),7.60-7.64(m, 5H, C₆H₅),7.91 (s, 1H, =CH), 9.52 (s, 1H, NH); ¹³C NMR(δ ppm):13.47(-CH₃), 108.48(=CH), 115.16(=C<), 124.21(2×=CH), 126.23(2×=CH), 127.58(=CH), 128.00(=CH), 128.43(2×=CH), 128.94(2×=CH), 137.17(=C<), 138.60(=CH), 139.71(=C<), 142.22(=C<), 144.40(=C<), 149.78 (=C<);MS :*m/z* (M⁺1 300).

3'-(4-Methoxy-phenyl)-5-methyl-2-phenyl-2H, 1'H-[3,4'] bipyrazolyl (5c):

Yield:78%; Time(min.): 6; m. p. 217^{0} C; m. f. $C_{20}H_{18}N_{4}O$; IR (KBr, v_{max} , cm⁻¹):1160.36(OCH₃), 1217.24(N-N), 1500.04(C=C), 1577.84(C=N), 3012.04(Ar-CH), 3300.56(NH); ¹HNMR (δ ppm): δ 2.36(s, 3H, CH₃), 4.00(s, 3H, OCH3), 6.44(s, 1H, =CH), 7.51-7.54 (m, 5H, C₆H₅), 7.71-7.74(m, 4H, C₆H₄), 7.89(s, 1H, =CH), 9.59(s, 1H, NH); ¹³C NMR(δ ppm):13.43(-CH₃), 55.34(OCH₃), 108.46(=CH), 114.23(2×CH), 115.10(=C<), 124.65(2×=CH), 127.34(=CH), 128.41(2×=CH), 128.94(2×=CH), 137.15(=C<), 138.83(=C<), 139.74(=C<), 142.35(-CH), 144.41(=C<), 149.48(=C<), 160.04(=C<); MS : *m/z* (M⁺1 330).

2-(2,4-Dinitro-phenyl)-3'-(4-methoxy-phenyl)-5-methyl-2H,1'H-[3,4']bipyrazolyl(5d): Yield:81%; Time(min.): 6; m. p. 229⁰C; m. f. C₂₀H₁₆N₆O₅; IR (KBr, v_{max}, cm⁻¹):1295(C-O), 1515(-NO₂), 1595(C=N), 1600 (C=C), 3040(Ar-H), 3333(-NH); ¹HNMR (δ ppm): δ

1515(-NO₂), 1595(C=N), 1600 (C=C), 3040(Ar-H), 3333(-NH); ¹HNMR (δ ppm): δ 2.33(3H,s,CH₃), 3.80(3H,s,OCH₃), 6.44(H,s,=CH), 6.97-7.73(4H,m,C₆H₄), 7.42(H,d,=CH), 7.93(H,d,=CH), 8.48(H,d,=CH), 8.78(H,s,=CH), 9.52(H,d,NH); ¹³C NMR(δ ppm): 13.44(–CH₃), 55.37(OCH₃), 108.47(=CH), 114.21(2×=CH),115.07(=C<), 121.22(=CH), 128.22(=CH), 128.40(2×=CH), 130.13(=CH), 136.81(=C<), 137.14(=C<), 138.84(=CH), 142.38(=C<), 144.42(=C<), 145.41(=C<), 146.51(=C<), 149.50(=C<), 160.07(=C<); MS : m/z (M⁺1 496).

3'-(4-Methoxy-phenyl)-5-methyl-2,1'-diphenyl-2H,1'H-[3,4']bipyrazolyl (5e): Yield:79%; Time(min.): 6; m. p. 212^{0} C; m. f C₂₆H₂₂N₄O; IR (KBr, v_{max}, cm⁻¹):1293(C-O), 1598(C=N), 1605 (C=C), 3042(Ar-H),; ¹HNMR (δ ppm): δ 2.36(3H,s,-CH₃), 3.82(3H,s,OCH₃), 6.44(H,s,=CH), 6.97-7.73(4H, m, C₆H₄), 7.11-7.31(5H,m,C₆H₅), 7.33-7.72(5H,m,C₆H₅), 8.27(H,s,=CH); ¹³C NMR(δ ppm): 13.45(-CH₃), 55.35(OCH₃), 108.48(=CH), 114.23(2×=CH), 115.09(=C<), 118.95(2×=CH), 124.36(=CH), 124.68(2×=CH), 126.85 (=C<), 127.36 (=C<), 128.41(2×=CH), 128.96(2×=CH), 129.50(2×=CH), 137.17(=C<), 139.71(=C<), 139.90 (=C<), 142.37(=C<), 144.40 (=C<), 149.49(=C<), 160.05(=C<); MS : *m/z* (M⁺1 406).

2-(2,4-Dinitro-phenyl)-3'-(4-methoxy-phenyl)-5-methyl-1'-phenyl-2H,1'H-

[3,4']bipyrazolyl (5f): Yield:83%; Time(min.): 6; m. p. 241^{0} C; m. f. $C_{26}H_{20}N_{6}O_{5}$; IR (KBr, v_{max} , cm⁻¹):1288(C-O), 1515 (-NO₂), 1583(C=N), 1585 (C=C), 3045(Ar-CH); ¹HNMR (δ ppm): δ 2.37(3H,s,-CH₃), 3.82(3H,s,OCH₃), 6.44 (H,s,=CH),6.96-7.72(4H,m, C₆H₄), 7.31-

7.70(5H,m,C₆H₅), 7.41(H,d,=CH), 8.26(H,s,=CH),8.47(H,d,=CH) ,8.79 (H, s, =CH); ¹³C NMR(δ ppm): 13.43(-CH₃), 55.36(OCH₃), 108.50(=CH), 114.24(2×=CH), 115.11(=C<) , 118.96(2×=CH), 121.21(=CH), 124.37 (=CH), 126.86(=C<), 128.21(=CH), 128.42(2×=CH), 129.51(2×=CH), 130.11(=CH), 136.81(=C<),137.18(=C<), 139.91(=C<),142.36(=C<),144.41(=C<),145.41(=C<),146.51(=C<),149.51(=C<),160.06(=C<); MS : *m/z* (M⁺1 496).

5-Methyl-3'-(4-nitro-phenyl)-2-phenyl-2H,1'H-[3,4']bipyrazolyl (5g):

Yield:81%; Time(min.): 7; m. p. 249^{0} C; m. f. $C_{19}H_{15}N_{5}O_{2}$; IR (KBr, v_{max} , cm⁻¹):1062.78(C-N), 1217.08(N-N), 1421.54(-NO₂), 1589.34(C=N),1614.42(C=C), 3095.75(Ar-CH), 3309.85(NH) ;¹HNMR (δ ppm): 2.38(s, 3H, CH₃), 6.46(s, 1H, =CH),7.41-7.44(m, 5H, C₆H₅),7.90(s, 1H, =CH), 8.21-8.24(m, 4H, C₆H₄), 9.56(s, 1H,NH); ¹³C NMR(δ ppm): 13.42(-CH₃), 108.47(=CH), 115.09(=C<),123.88(2×=CH), 124.68(2×=CH), 127.32(=CH), 127.93(2×=CH), 128.94(2×=CH), 137.17(=C<), 138.79(=CH), 139.70(=C<), 142.37(=C<), 144.39(=C<), 147.49(=C<), 149.52 (=C<); MS : *m/z* (M⁺1 345).

2-(2,4-Dinitro-phenyl)-5-methyl-3'-(4-nitro-phenyl)-2H,1'H-[3,4']bipyrazolyl(5h):

Yield:76%; Time(min.): 6; m. p. 239⁰C; m. f. $C_{19}H_{13}N_7O_6$; IR (KBr, v_{max} , cm⁻¹):1594 (C=C), 1520(-NO₂), 1598(C=N), 3033 (Ar-CH), 3330 (-NH); ¹HNMR (δ ppm): δ 2.34(3H,s,-CH₃), 6.44(H,s,=CH), 7.41(H, d, =CH), 7.80-8.24(4H, m, C₆H₄), 7.90(H,d,=CH), 8.47(H,d,=CH), 8.77(H,s,=CH) 9.51(H,d,NH),; ¹³C NMR(δ ppm): 13.44(-CH₃), 108.47(=CH), 115.07(=C<), 121.21(=CH), 123.96(2×=CH), 127.92(2×=CH), 128.21(=CH), 130.11(=CH), 136.81(=C<), 137.19(=C<), 138.83(=CH), 142.36(=C<), 144.42 (=C<), 145.02(=C<), 146.52(=C<), 147.50 (=C<), 149.48(=C<); MS : *m/z* (M⁺1 435).

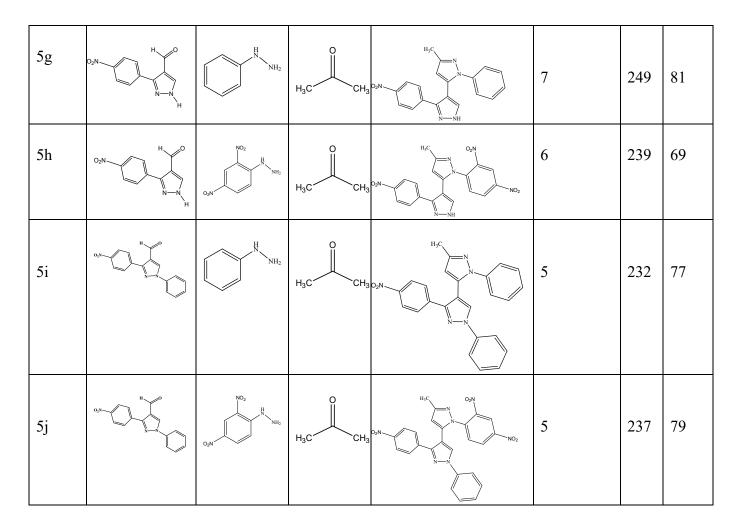
5-Methyl-3'-(4-nitro-phenyl)-2,1'-diphenyl-2H,1'H-[3,4']bipyrazolyl(5i):Yield:77%;

Time(min.): 6; m. p. 232^{0} C; m. f. $C_{25}H_{19}N_{5}O_{2}$; IR (KBr, v_{max} , cm⁻¹):1513(-NO₂), 1595 (C=N), 1600 (C=C), 3040 (Ar-CH); ¹HNMR (δ ppm): δ 2.36(3H,s,-CH₃), 6.45(H,s,=CH), 7.11-7.31(5H, m, C₆H₅), 7.33-7.72(5H,m,C₆H₅), 7.80-8.24(4H,m,C₆H₄),8.27(H,s,=CH); ¹³C NMR(δ ppm): 13.43(–CH₃), 108.49(=CH), 115.09(=C<), 118.95(2×=CH), 121.20(=CH), 123.94(2×=CH), 124.36(=CH), 126.85 (=C<), 127.91(2×=CH), 128.20(=CH), 129.50(2×=CH), 130.10(=CH), 136.80(=C<), 137.17(=C<), 139.90 (=C<), 142.37(=C<), 144.40 (=C<), 145.00(=C<)146.50(=C<), 147.49 (=C<), 149.50(=C<);MS : *m/z* (M⁺1 421).

2-(2,4-Dinitro-phenyl)-5-methyl-3'-(4-nitro-phenyl)-1'-phenyl-2H,1'H-[3,4']bipyrazolyl

(5j):Yield:79%; Time(min.):6; m. p. 237^{0} C; m. f. $C_{25}H_{17}N_{7}O_{6}$; IR (KBr, v_{max} , cm⁻¹):1525(-NO₂), 1610 (C=C), 1635 (C=N), 3050 (Ar-CH); ¹HNMR (δ ppm): δ 2.34(3H,s,-CH₃), 6.47(H,s,=CH), 7.33-7.72(5H,m,C₆H₅), 7.80-8.24(4H,m,C₆H₄), 7.42(H,d,=CH), 8.26(H,s,=CH), 8.48(H,d,=CH), 8.78(H,s,=CH); ¹³C NMR(δ ppm): 13.42(-CH₃), 108.44(=CH), 115.06(=C<), 118.95(2×=CH), 121.22(=CH), 123.94(2×=CH),124.34(=CH), 126.85 (=C<), 127.93(2×=CH), 128.25(=CH), 129.50(2×=CH), 130.12(=CH),136.80(=C<), 137.16(=C<), 139.91 (=C<), 142.36(=C<), 144.43 (=C<), 145.03(=C<) 146.51(=C<), 147.49 (=C<),149.52(=C<);MS : *m/z* (M⁺1 511).

TABLE-1 PREPARATION OF BIPYRAZOLYL UNDER MICROWAVE IRRADIATION								
Comp.	Substrate			Product	Time (min.)	m.p. (⁰ C)	Yield (%)	
	1	2	3					
5a		O2N NO2	H ₃ C CH ₃		5	215	84	
5Ъ		NH ₂	H ₃ C CH ₃	H ₃ C N N N N NH	5	209	74	
5c	H ₃ CO H ₃ CO H ₁ CO H	NH ₂	H ₃ C CH ₃	H ₃ CO	5	215	78	
5d		O2N NO2 H NH2	H ₃ C CH ₃	$H_{3}CO$ $()$ $H_{1}C$ $O_{2}N$ $O_{2}N$ $H_{3}CO$ $()$ NO_{2} NO_{2} NO_{2}	6	229	81	
5e	$H_{\mathcal{L}}(O) \longrightarrow \left(\int_{\mathcal{L}} \int_$	NH ₂	н ₃ с сн ₃	H ₃ CO	5	212	79	
5f	и,со-()+()	NO2 NH2	H ₃ C CH ₃	$H_3CO \rightarrow \begin{pmatrix} H_3C \\ H_3CO \end{pmatrix} \rightarrow \begin{pmatrix} H_3CO \\ $	6	241	83	



III. Result and discussion

- 1. The substances [3(a-e)] to [5(a-j)] display the range of potential biological actions.
- 2. From virtually all the compounds [3(a-e) to [5(a-j)], only two activities, namely HMGCS2 expression enhancer and Liver fibrosis therapy, are expected.
- 3. The compounds [5-g] and [5-i] are anticipated to exhibit spectacular HMGCS2 expression enhancer activity. Additionally, because of the phenyl ring, [5-g] is anticipated to be more active than the monocyclic molecule [5-i].
- 4. Each of the compounds [3(a-e)] and [5(a-j)] is anticipated to have therapeutic potential for liver fibrosis. But out of all the compounds, [5-b] is probably the most active.
- 5. The bipyrazolyl compound [5-b] is likewise anticipated to have spectacular signal transduction pathways inhibitor activity. This is more likely to have substrate activity for the cyclin-dependent kinase 2 inhibitor.
- 6. It is also anticipated that some of the drugs in this series will have MAP3K5 inhibitor, hepatic disorders therapy, CDK9/cyclin T1 inhibitor, UGT2B12 substrate, Cyclindependent kinase inhibitor, and CYP19A1 expression inhibitor properties.

7. Due to the differences from the known pharmacological drugs, several of the compounds have Pa values between 0.5 and 0.7, making it conceivable for them to demonstrate activities. They are also most likely new chemical entities.

The substances [5-g] and [5-i] are proposed for the expression enhancer activity of the gene HMGCS2, which controls the synthesis of ketone bodies in the liver and extrahepatic organs.

QSAR Analysis of Activities with PASS:

A number of biological functions were examined in relation to structure using the computer programme PASS. Using the structures of derivatives [5(a-j)], software was used to forecast the probabilities of being active [Pa] and inactive [Pi] given a set of biological activities. The following three activities were projected to have the highest probabilities for the chemical series [5(a-j)].

i) HMGCS2 expression enhancer(ii) MAP3K5 inhibitor(iii) Liver fibrosis treatment.

(i) HMGCS2 expression enhancer:

The gene HMGCS2, which regulates ketone body synthesis, is found in the liver and several extrahepatic tissues, including the colon. This gene's expression alters when the CaCo-2 colonic epithelial cells differentiate. HMGCS2 is mostly detected in differentiated cells of the human colonic epithelium, according to immunohistochemistry utilising specific antibodies. The test indicates that HMGCS2 is a direct target of c-Myc, which contains HMGCS2 transcriptional activity. Blocking Miz-1's trans activating activity, which often takes place through a Sp1-binding site in the gene's proximal promoter, mediates c-Myc trans repression. 90% of Myc-dependent colon and rectum tumours had decreased expression of the human HMGCS2 gene.HMGCS2 protein expression is down-regulated in carcinomas with moderate and poor differentiation. Additionally, it is down-regulated in 80% of small intestinal tumours without Myc.

(ii) MAP3K5 inhibitor:

288 melanomas with the MAP3K5 popularity screen were still present in the harbour. All MAP3K5-mutated samples included the wild type of BRAF, proving that the two mutations are mutually exclusive. According to a functional analysis of the MAP3K5 R256C mutation, the decreased MKK4 activation caused by increased binding of the inhibitory protein thioredoxin led to higher proliferation and anchorage-independent growth of melanoma cells. This mutation indicates a possible target for the development of novel melanoma treatments.

(iii)Liver fibrosis treatment:

As a result of an imbalance between fibrous hyperplasia and fibrous breakdown, liver fibrosis is the excessive loss of fibrous connective tissue in the liver. The body's response to damage healing is called fibrous hyperplasia. Continuous fibrous hyperplasia of the liver and the development of liver fibrosis can result from repeated chronic liver parenchyma necrosis and inflammation, which can be caused by a variety of factors. Liver fibrosis is a necessary stage for chronic liver disorders to progress into liver cirrhosis from the pathological alterations of various chronic liver diseases. It is stated that while liver cirrhosis cannot be reversed, hepatic fibrosis can.

Table 2								
PREDICTIONS OF BIOLOGICAL ACTIVITIES BY PASS								
	HMGCS2 expression	MAP3K5	Liver fibrosis treatment					
Comp.	enhancer	inhibitor						
	Pa	Pa	Pa					
5a	0.606	0.498	0.438					
5b	0.606	0.576	0.605					
5c	0.539	0.521	0.442					
5d	0.553	0.459	0.358					
5e	0.511	0.513	0.250					
5f	0.577	0.434	0.207					
5g	0.702	0.516	0.472					
5h	0.609	0.501	0.448					
5i	0.694	0.507	0.260					
5j	0.636	0.478	0.252					

IV. Conclusion

As a result, we have developed a quick and efficient method that might be used in the creation of new drugs for the selective synthesis of new 5-methyl-2, 1, 3'-triphenyl-2H, 1'H-[3, 4'] bipyrazolyl in an aqueous medium.

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References

- 1. L. Claisen, P. Roosen, Justus Liebigs Ann. Chem., (1893), 278-274.
- L.C. Behr, R. Fusco, C.H. Jarboe, The Chemistry Of Heterocyclic Compounds. Vol. 22.Pyrazoles, Pyrazolines, Pyrazolidines, Indazoles And Condensed Rings., Wiley R. H.Ed., John Wiley & Sons, New York, (1967).

3. S.A.F. Rostom, PolysubstitutedPyrazoles, Part 6. Synthesis Of Some 1-(4-Chlorophenyl)-4-Hydroxy-1h-Pyrazol-3-Carbonyl Derivatives Linked To NitrogenousHeterocyclic Ring Systems As Potential Antitumor Agents, Bioorg. Med. Chem., 18(7) (2010), 2767-2776.

4. B. Haresh,O.Dharti, G. Joshi, H. Parekh.Synthesis Of Some Novel Pyrazolines As Biologically Potent Agents, 3(3) (1997),239-244.

- A. Cuadro, J. Elguero, P. Navarro.Binuclear Pyrazoles. I. Synthesis And Cytotoxic Activity Of 1,1'-Dibenzyl And 1,1'-Dihydroxymethyl 4,4'-Bispyrazoles, Chem. Pharm. Bull., 33(6)(1985), 2535-2540.
- I. Bouabdallah, L.A.M'ebarek, A. Zyad, A. Ramdani, I. Zidane, A.Melhaoui, Pyrazolic Compounds As Cytotoxic Agents., Nat. Prod. Res., 21 (2007).298-302.

7. V. Kapase ,Microwave Assisted Synthesis and QSAR study of 1-substituted-3-aryl-

- 1H-pyrazole-4- carbaldehydes derivatives, Asian Journal of Organic & Medicinal Chemistry,6(2) (2021), 79-83.
- P. T. Chovatia, J. D. Akabari, P. K. Kachhadia, P. D. Zalavadia , H. S. Joshi.Synthesis And Selective Antitubercular And Antimicrobial Inhibitory Activity Of 1-Acetyl-3,5-Diphenyl-4,5-Dihydro-(1h)-Pyrazole Derivatives, J. Serb. Chem. Soc. 71 (7) (2007), 713–720.
- S. Kalepu, N.Meghana, S.Nichitha, K. Bhavani, A. Yadhav, M.Pranavi , Microwave Assisted Synthesis And Antimicrobial Evaluation Studies On Pyrazolines. Jpbs, 8(3)(2018), 195-199.
- 10. S. Khot, V. Kapase, S. Kenwade, S. Dhongade, Microwave Assisted Multicomponent Synthesis of Promising Insulin Inhibitor and Mcl-1 Antagonist Thiazolidinone&PyrazoloThiazolidine Derivatives, IN Jr. Pharmacy Research Scholars, , 3(1) (2014) 363-373.
- 11. U. Sahoo, S. Basak, R. Chawla. Microwave Assisted Synthesis, Characterization And Antimicrobial Activity Of Novel Bipyrazole Derivatives, Pharmawave,10 (2017),40-47.
- S. Dhongade-Desai, V. Kapase.Synthesis, Characterization And Biological Prediction Study Of Pyrazol-1-Yl-Thiazolidin-4-One Derivatives, World Journal Of Pharmacy And Pharmaceutical Sciences ,5(7) (2016), 923-931.
- 13. A.Nayak, A.S.Mittra.4,4-Bis-5-Pyrazolones And Their 4,4-Unsaturated Products For Possible Use As Fungicides, J. Ind. Chem. Soc., 57 (1980), 643-649.
- S. Dhongade , V.Kapase, S. Khot , Microwave Assisted Synthesis & QSAR Study of Some Novel Pyrazolethioamide Derivatives, IN Jr. Pharmacy Research Scholars, 3(1) (2014), 243-250.

15. I. Pearson. Us.Pat, 3883549, Chem. Abstr., 84, 81243(1975).

- T.Igarashi, K.Sakurai, T.Oi.,H.Obara, H.Ohya, H.Kamada.New Sensitive Agents For Detecting Singlet Oxygen By Electron Spin Resonance Spectroscopy Free Radical Biol. Med., 26(10) (1999),1339-1345.
- F. Bakr A. Wahab, K. Dawood.Synthesis And Applications Of Bipyrazole Systems, Arkivoc, 1(2012), 491-545.

18. A. Padwa (Ed.).1.3-Dipolar Cycloaddition Chemistry. Vol.1., Wiley: New York, 1984.

 B. Sun, K. Adachi, M. Noguchi.Intramolecular 1,3-Dipolar Cycloaddition At The Periphery Of Heterocyclic Systems. Part 3. A Facile Hydrazone-Azomethine Imine Isomerization At The Periphery Of Pyridine And Pyrido[1,2-A]Pyrimidine Systems, Tetrahedron, 52(3) (1996), 901-914.

20. R.Grigg. Prototropic Routes To 1,3- And 1,5-Dipoles, And 1,2-Ylides: Applications To The Synthesis Of Heterocyclic Compounds, Chem. Soc. Rev., 16 (1987), 89-121.